

NEXCERIS where energy meets environment

Supircat<sup>TM</sup>: A high performing internal reforming catalyst technology for MS&T18, COLUMBUS, OH SOFC applications



- Brief Company introduction
- Focus of today's presentation
- Background
- Reactor design and testing system
- Internal reforming results
- Other studies
- Conclusions







- ▶ Founded in 1994, privately held
- Technology Developer
  - advanced ceramics, electrochemical devices
- Product Developer
  - sensors and monitors, solid oxide fuel cells, catalysts
- Manufacturer/Distributor
  - sensors, solid oxide fuel cells and related products
- ISO 9001:2015 certification
  - covers all products and operations







**fuelcell**materials.com is our sales division to supplying high quality fuel cell and battery materials, coatings, and related materials for R&D and OEM markets.











#### **Powders & Pastes**

• Standard and custom powders for SOFC researchers



## Cells and Substrates

• Many sizes, formulations and geometries for reproducible results



## Test Fixtures and Systems

• For researchers needing credible data



## Components and Other

• Seals, process materials, current collecting meshes, etc.







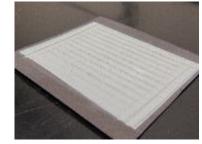
# Focus of today's talk





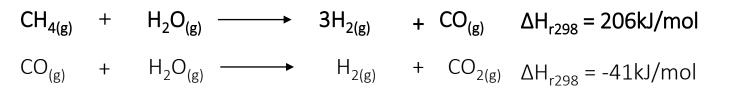
Powder





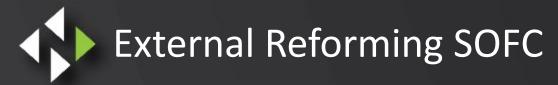
Coated on High  $\kappa$  Part

Cell Fuel System

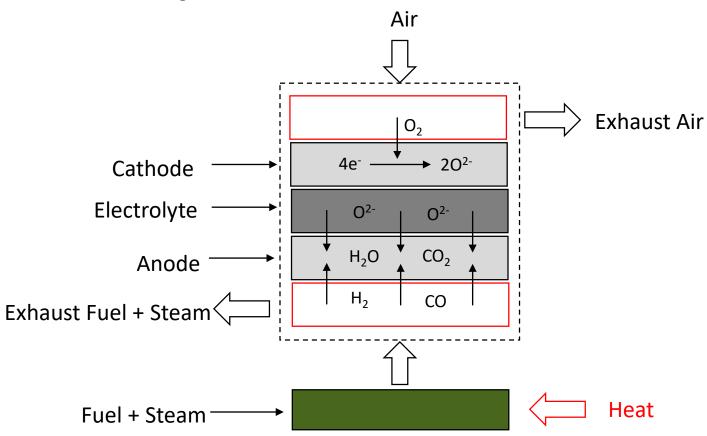








# **Principles of ER SOFC**



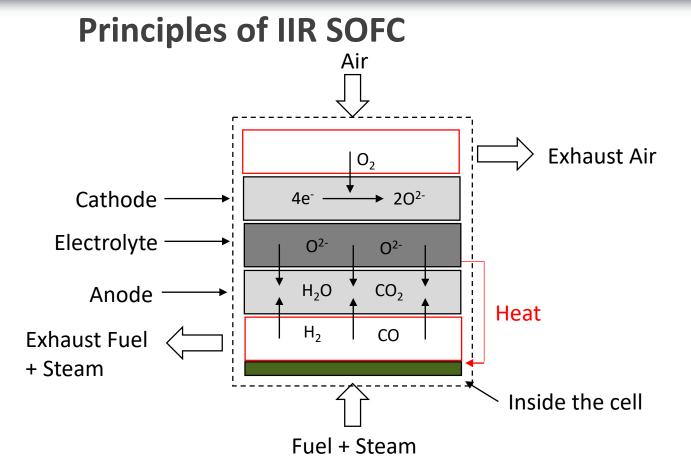
Endothermic reforming reaction takes place outside the cell

Heat for reforming is provided from an external source

Complex cell design



Indirect Internal Reforming SOFC



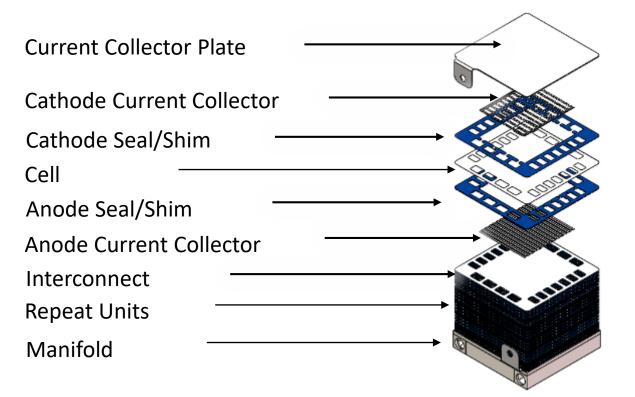
Endothermic fuel reforming reaction takes place inside the cell (close to anode)

- Heat for reforming is harnessed from exothermic
  - electrochemical reaction
- Simplified cell design (more attractive)





# **SOFC Stack Repeat Unit**



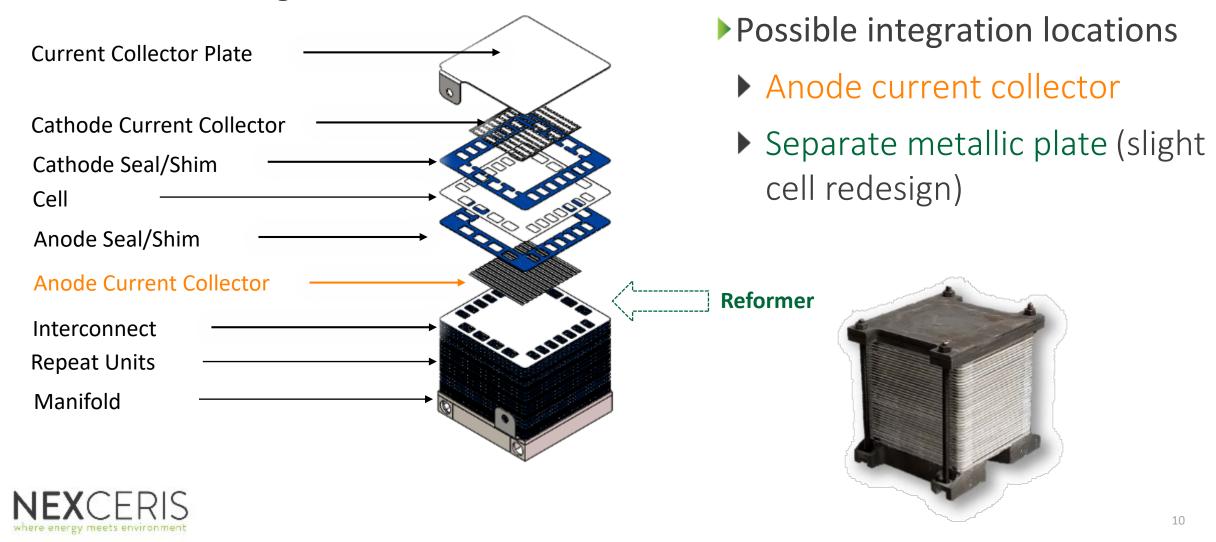
U.S. Patent No. 8,968,956

- FlexCells: ScSZ electrolyte material
- Thin-foil interconnects: Crofer 22 APU with cathode-face coatings
- Seals: Glass ink and ceramic/glass composites
- Shims: Alloys or inorganic materials
- Cathode current collectors: Coated metal alloy meshes
- Anode current collectors: Nickel foam, coatings to preserve sulfur tolerance



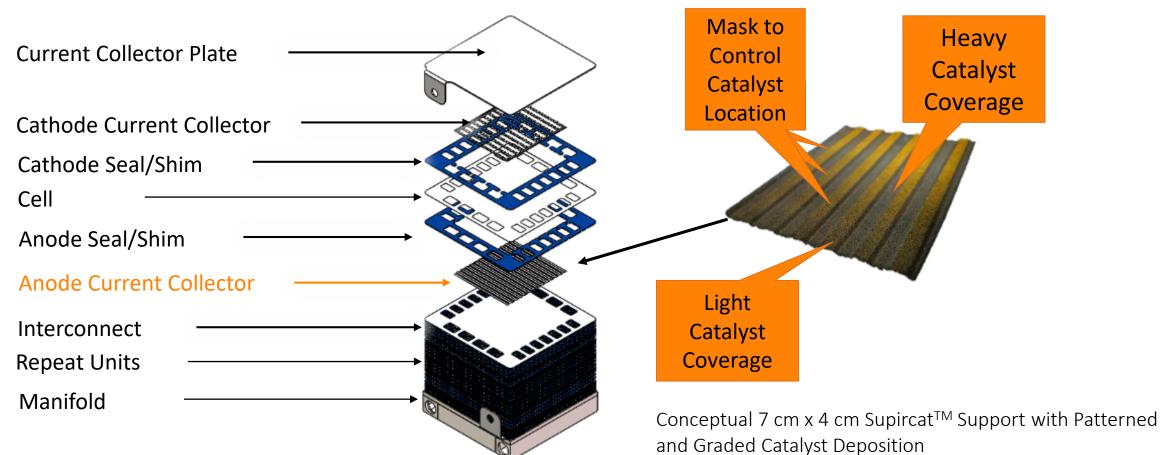


Where to integrate reformer





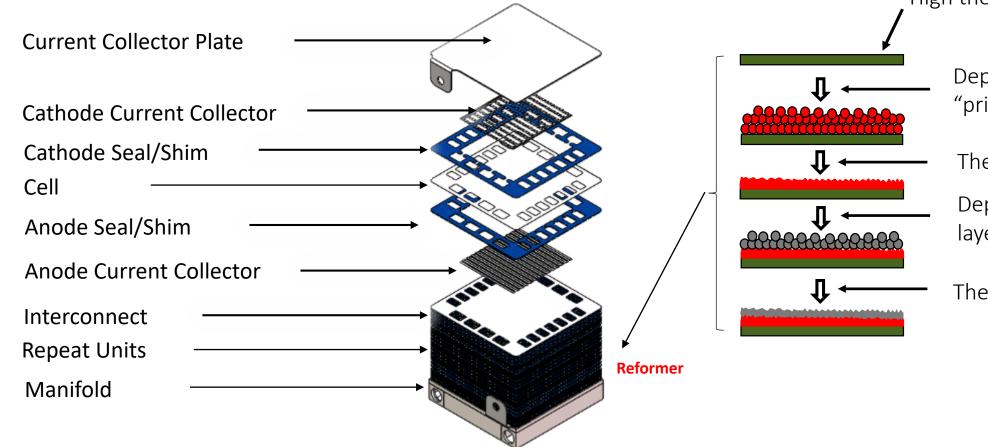
## Use of Ni foam current collector







## Use of high thermal conductivity substrate





High thermal conductivity substrate

Deposition of a protective "primer" layer (Alumilok<sup>™</sup>)

Thermal treatment

Deposition of a catalyst layer (Supircat<sup>TM</sup>)

Thermal treatment



#### **Catalyst Powder**



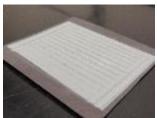
- Cost
- Immediate activation (no reduction needed)
- Equilibrium activity
- Sulfur tolerant
- Resistant to coking
- Performance stability

#### Slurry Suspension



- Optimized flow characteristics (rheological properties)
- Optimized solids loading
- Spray-ability

#### Coating



- Controlled layer thickness
- Layer adhesion





## **Spraying Capabilities**

# Manual spraying



# Automated spraying

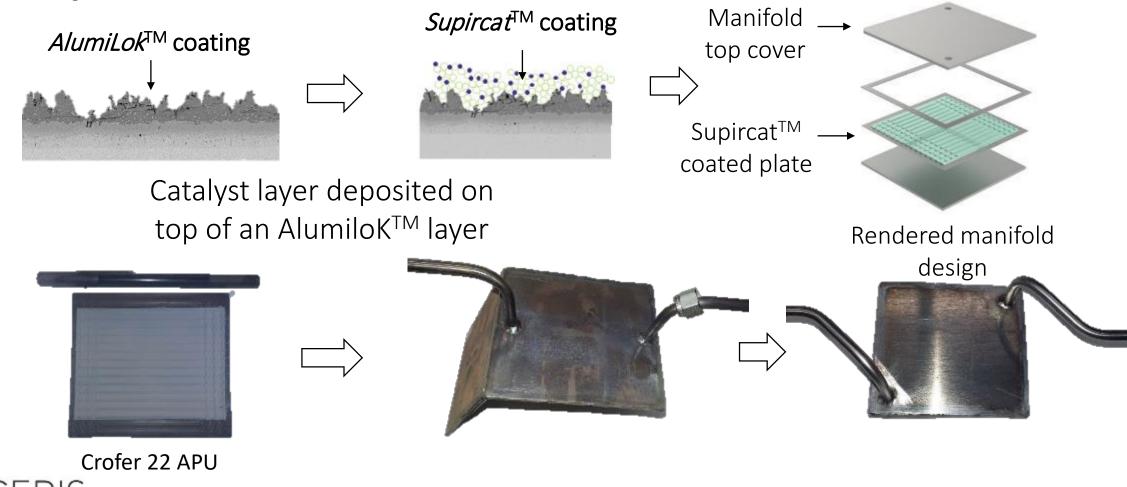


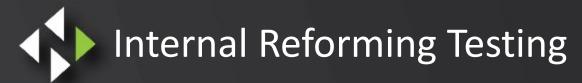




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# **Compact manifold**





# **Testing & Analysis System and Conditions**



Test rig

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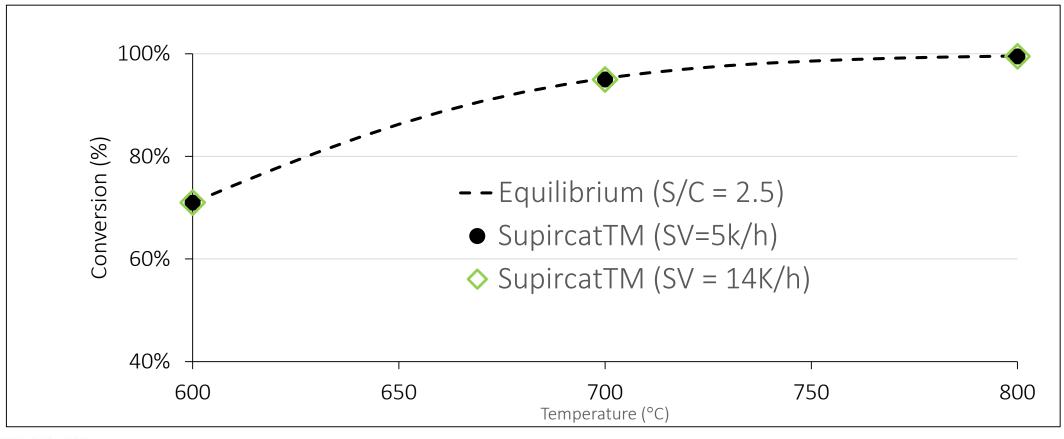


Analysis system

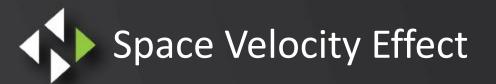
Table 1. Testing Conditions		
Parameter	Min.	Max.
Temperature (°C)	500	850
Pressure (bara)	1	1.5
GHSV (h <sup>-1</sup> )	3,000	35,000
S:C ratio	2.5	4.0



## Conditions: Temp: 600 – 800 °C, Pres: 1 bara, GHSV: 5 & 14K, S/C: 2.5

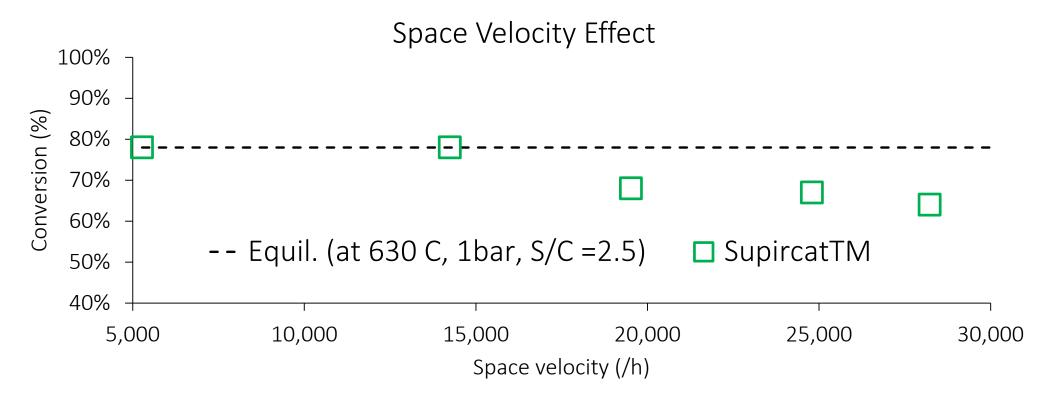




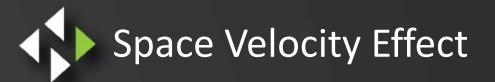


# At isothermal conditions

Conditions: Temp: 630 °C, Pres: 1 bara, S/C: 2.5

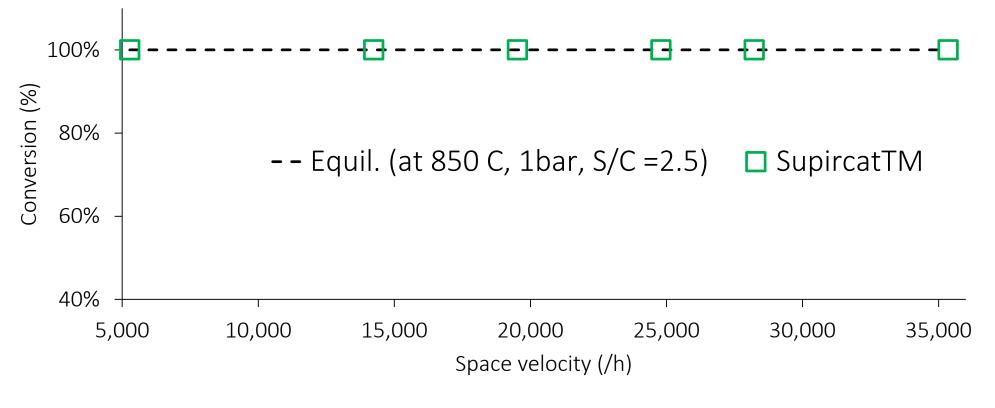






# **Isothermal Conditions**

Conditions: Temp: 850 °C, Pres: 1 bara, S/C: 2.5

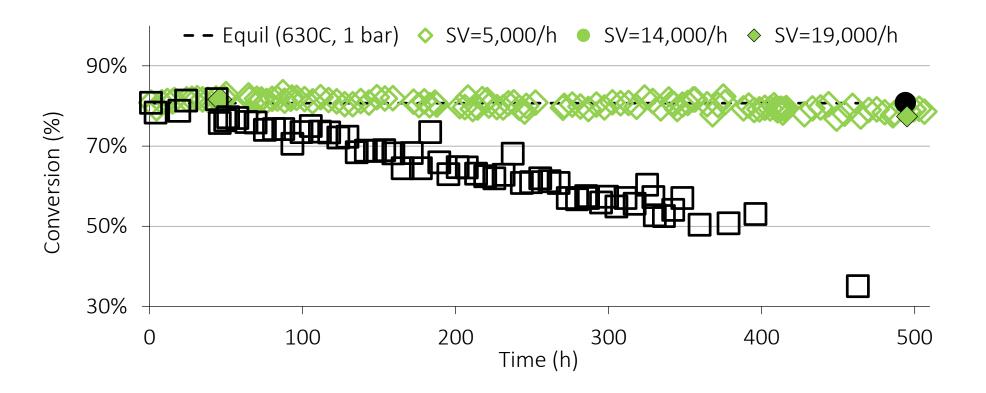






# **Effect of Poison**

## Conditions: Temp: 850 °C, Pres: 1 bara, S/C: 2.5







- ► Alumilok<sup>TM</sup> and Supircat<sup>TM</sup> catalyst technologies proven active
  - Baseline performance
  - ► Life-time stability
- Good tolerance to poisons e.g. Cr and S



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